# metal-organic papers

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## Jinglin Chen, Lei Han and Zhongning Chen\*

State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Fuzhou, Fujian 350002, People's Republic of China

Correspondence e-mail: czn@ms.fjirsm.ac.cn

#### **Key indicators**

Single-crystal X-ray study T = 293 KMean  $\sigma(\text{C-C}) = 0.008 \text{ Å}$  R factor = 0.047 wR factor = 0.117 Data-to-parameter ratio = 16.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# *trans*-Bis(isothiocyanato)tetrapyridineruthenium(II) dichloromethane disolvate

The title compound, *trans*-[Ru(NCS)<sub>2</sub>( $C_5H_5N$ )<sub>4</sub>]·2CH<sub>2</sub>Cl<sub>2</sub>, was prepared by the reaction between *trans*-dichlorotetrapyridine-ruthenium(II) and excess KSCN in refluxing aqueous pyridine. The Ru atom is in a pseudo-octahedral environment, with two N-donors from two monodentate NCS groups and four N-donors from the pyridine ligands. The complex molecule lies on a crystallographic twofold rotation axis, passing through Ru and the two pyridine ligands

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#### Comment

The crystal structure of the title compound, (I), consists of discrete trans-[Ru(NCS)<sub>2</sub>(C<sub>5</sub>H<sub>5</sub>N)<sub>4</sub>] molecules and dichloromethane solvent molecules. The Ru atom is coordinated octahedrally by four N atoms from the pyridine ligands and by two N atoms from the monodentate NCS groups in an octahedral arrangement. The  $Ru-N(C_5H_5N)$  lengths are in the range 2.072 (5)-2.096 (5) Å, close to those in the previously reported compounds trans-[Ru(CN)<sub>2</sub>(C<sub>5</sub>H<sub>5</sub>N)<sub>4</sub>]·2MeCN (Coe et al., 1995a) and trans-[RuCl(C<sub>5</sub>H<sub>5</sub>N)<sub>4</sub>(PhCN)]PF<sub>6</sub> (Coe et al., 1995b). The NCS groups are essentially linear. The N4-C1 [1.159 (5) Å] and C1-S1 [1.632 (4) Å] lengths indicate tripleand single-bond character, respectively, and are similar to those in the reported complexes [Ni(NCS)<sub>2</sub>(C<sub>5</sub>H<sub>5</sub>N)<sub>4</sub>] (Valach et al., 1984) and  $[Ni(en)_2(NCS)_2] \cdot C_6 H_6$  (Squattrito et al., 1996). The C-C [1.339(9)-1.389(7) Å] and C-N [1.323(6)-1.346 (5) Å] lengths are in the normal ranges for pyridine ligands (Evans et al., 1973; Coe et al., 1995a,b). The mean bond angles in the pyridine ligands  $[C-N-C = 116.2 (7)^{\circ}, N-C C = 123.4 (8)^{\circ}$ , and C-C-C = 118.4 (4) and  $119.6 (4)^{\circ}$  are similar to those observed in  $[Ni(NCS)_2(py)_4]$  [C-N-C =  $116.8 (5)^{\circ}$ , N-C-C = 122.8 (5)°, and C-C-C = 117.1 (6) and 120.2 (6)°; Valach et al., 1984].



## Experimental

© 2002 International Union of Crystallography Printed in Great Britain – all rights reserved The title compound was synthesized by the reaction of a pyridine solution of *trans*-[RuCl<sub>2</sub>( $C_5H_5N$ )<sub>4</sub>] (Evans *et al.*, 1973) with an excess of an aqueous solution of KSCN. After refluxing for 1 h, the solution was put aside at room temperature to give a yellow precipitate. The

solid was then filtered off, washed with methanol and ether, and dried in air. Well-shaped crystals were grown by slow diffusion of hexane into a dichloromethane solution at room temperature.

 $D_x = 1.474 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation Cell parameters from 3186

reflections

 $\theta = 2.0-25.0^{\circ}$  $\mu=0.99~\mathrm{mm}^{-1}$ 

T = 293 (2) K

Prism, yellow

 $0.40 \times 0.35 \times 0.28 \text{ mm}$ 

#### Crystal data

[Ru(NCS)2(C5H5N)4]·2CH2Cl2  $M_r = 703.48$ Monoclinic, C2/ca = 12.9410 (8) Å b = 16.1308 (11) Åc = 15.4715 (11) Å $\beta = 100.988 \ (2)^{\circ}$  $V = 3170.4 (4) \text{ Å}^3$ Z = 4Data collection Siemens SMART CCD diffractometer  $\omega$  scans

2765 independent reflections 2352 reflections with  $I > 2\sigma(I)$  $R_{\rm int}=0.028$  $\theta_{\rm max} = 25.0^\circ$ Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  $h = -14 \rightarrow 15$  $k = -19 \rightarrow 10$  $T_{\rm min}=0.681,\ T_{\rm max}=0.759$  $l=-13\rightarrow 18$ 

#### Refinement

4923 measured reflections

Refinement on  $F^2$  $w = 1/[\sigma^2(F_o^2) + (0.0355P)^2]$  $R[F^2 > 2\sigma(F^2)] = 0.047$ + 11.5869P]  $wR(F^2) = 0.117$ where  $P = (\dot{F_o}^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\rm max} < 0.001$ S = 1.15-3 2765 reflections  $\Delta \rho_{\rm max} = 0.51 \text{ e} \text{ Å}$  $\Delta \rho_{\rm min} = -0.86 \text{ e } \text{\AA}^{-3}$ 170 parameters H-atom parameters constrained

#### Table 1

Selected geometric parameters (Å, °).

Ru1-N4	2.025 (3)	Ru1-N1	2.096 (5)
Ru1–N3	2.072 (5)	S1-C1	1.632 (4)
Ru1–N2	2.082 (3)	N4-C1	1.159 (5)
N4 <sup>i</sup> -Ru1-N4	179.8 (2)	N2 <sup>i</sup> -Ru1-N2	179.7 (2)
N4-Ru1-N3	89.88 (11)	N4-Ru1-N1	90.12 (11)
N4 <sup>i</sup> -Ru1-N2 <sup>i</sup>	88.84 (13)	N2-Ru1-N1	89.84 (11)
N4-Ru1-N2 <sup>i</sup>	91.16 (13)	C1-N4-Ru1	175.6 (3)
N4-Ru1-N2	88.84 (13)	N4-C1-S1	179.9 (5)
N3-Ru1-N2	90.16 (11)		

Symmetry code: (i) 1 - x, y,  $\frac{3}{2} - z$ .

The positions of the H atoms were generated geometrically (C-H bond fixed at 0.96 Å), assigned isotropic displacement parameters, and allowed to ride on their respective parent C atoms.



### Figure 1

A view of the title complex with the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. [Symmetry code: (A)  $1 - x, y, \frac{3}{2} - z$ .]

Data collection: SMART (Bruker, 1997); cell refinement: SMART; data reduction: SHELXTL (Bruker, 1999); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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### References

- Bruker (1997). SMART. Version 5.054. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (1999). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Coe, B. J., Meyer, T. J. & White, P. T. (1995a). Inorg. Chem. 34, 593-602.
- Coe, B. J., Meyer, T. J. & White, P. T. (1995b). Inorg. Chem. 34, 3600-3609.
- Evans, I. P., Spencer, A. & Wilkinson, G. (1973). J. Chem. Soc. Dalton Trans. pp. 204-209.
- Squattrito, P. J., Iwamoto, T. & Nishikiori, S. (1996). Chem. Commun. pp. 2665-2666.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Valach, F., Sivy, P. & Koren, B. (1984). Acta Cryst. C40, 957-959.